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IS 10369 (1982): Ethion, Technical [FAD 1: Pesticides and Pesticides Residue Analysis]



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IS : 10369 - 1982  
( Reaffirmed 2004 )

*Indian Standard*  
SPECIFICATION FOR ETHION, TECHNICAL

UDC 632.951 ETHION



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MANAK BHAVAN, 9 BAHADUR SHAH ZAFAR MARG  
NEW DELHI 110002

# *Indian Standard*

## SPECIFICATION FOR ETHION, TECHNICAL

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**AMENDMENT NO. 1 JANUARY 2007**  
**TO**  
**IS 10369 : 1982 SPECIFICATION FOR ETHION,**  
**TECHNICAL**

(Page 4, clause 3.1) — Substitute 'The material shall be packed as per requirements given in IS 8190 (Part 2) : 1988<sup>\*</sup>' for 'The material shall be packed as per requirements given in IS : 8190 (Part II) - 1980<sup>\*</sup>' as applicable to packing of malathion, technical.'

(Page 4, footnote marked<sup>\*</sup>) — Substitute 'Requirements for packing of pesticides: Part 2 Liquid pesticides ( *second revision* )' for the existing.

(Page 5, clause 4.1) — Substitute 'Representative samples of the material shall be drawn as prescribed in IS 10946 : 1996<sup>†</sup>' for 'Representative samples of the material shall be drawn as prescribed in 'Indian Standard methods for sampling of pesticides and their formulations' (*under preparation*).

(Page 5, clause 4.1, Note) — Delete Note and its contents.

(Page 5) — Insert the following footnote at the end of page:

<sup>\*</sup>'Methods of sampling for technical grade pesticides.'

(FAD 1)

# Indian Standard

## SPECIFICATION FOR ETHION, TECHNICAL

### 0. FOREWORD

**0.1** This Indian Standard was adopted by the Indian Standards Institution on 10 November 1982, after the draft finalized by the Pest Control Sectional Committee had been approved by the Agricultural and Food Products Division Council and the Chemical Division Council.

**0.2** Ethion, technical is employed in the preparation of number of insecticidal and acaricidal formulations.

**0.3** Ethion is the accepted common name by the International Organization for Standardization (ISO) for *ss*-methylene *ooo* *o*-tetraethyl di(phosphorodithioate). The empirical and structural formulæ and the molecular mass of ethion are as given below:

<i>Empirical Formula</i>	<i>Structural Formula</i>	<i>Molecular Mass</i>
$C_9H_{12}O_4P_2S_4$		384.5

**0.4** In the preparation of this standard, due consideration has been given to the provisions of the *Insecticides Act*, 1968, and the Rules framed thereunder. However, this standard is subject to the restrictions imposed under these, wherever applicable.

**0.5** For the purpose of deciding whether a particular requirement of this standard is complied with, the final value, observed or calculated, expressing the result of a test or analysis, shall be rounded off in accordance with IS:2-1960\*. The number of significant places retained in the rounded off value should be the same as that of the specified value in this standard.

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### 1. SCOPE

**1.1** This standard prescribes the requirements and methods of sampling and test for ethion, technical.

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\*Rules for rounding off numerical values (*revised*).



## 2. REQUIREMENTS

**2.1 Description** — The material shall be an amber liquid, free from extraneous impurities or added modifying agents.

**2.2** The material shall also comply with the requirements specified in Table 1.

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**TABLE 1 REQUIREMENTS FOR ETHION, TECHNICAL**

SL No.	CHARACTERISTIC	REQUIREMENT	METHOD OF TEST, REF TO	
			Appendix	Cl No. of IS: 6940-1982*
(1)	(2)	(3)	(4)	(5)
i)	Ethion content, percent by mass, <i>Min</i>	92.0	A	—
ii)	Acidity ( as $H_2SO_4$ ), percent by mass, <i>Max</i>	0.30	—	11.3
iii)	Moisture content, percent by mass, <i>Max</i>	0.20	—	4

\*Methods of tests for pesticides and their formulations ( *first revision* ).

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## 3. PACKING AND MARKING

**3.1 Packing** — The material shall be packed as per requirements given in IS: 8190 ( Part II )-1980\* as applicable to packing of malathion technical.

**3.2 Marking** — The containers shall bear legibly and indelibly the following information in addition to any other information as is necessary under the *Insecticides Act* and Rules framed thereunder:

- a) Name of the material;
- b) Name of the manufacturer;
- c) Date of manufacture;
- d) Batch number;
- e) Ethion content, percent ( *m/m* );
- f) Net mass of the contents; and
- g) A cautionary notice as worded in *Insecticides Act* and Rules.

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\*Requirements for packing of pesticides, Part II Liquid pesticides ( *first revision* ).

### 3.2.1 Each container may also be marked with the ISI Certification Mark.

**NOTE** — The use of the ISI Certification Mark is governed by the provisions of the Indian Standards Institution ( Certification Marks ) Act and the Rules and Regulations made thereunder. The ISI Mark on products covered by an Indian Standard conveys the assurance that they have been produced to comply with the requirements of that standard under a well-defined system of inspection, testing and quality control which is devised and supervised by ISI and operated by the producer. ISI marked products are also continuously checked by ISI for conformity to that standard as a further safeguard. Details of conditions under which a licence for the use of the ISI Certification Mark may be granted to manufacturers or processors, may be obtained from the Indian Standards Institution.

## 4. SAMPLING

**4.1** Representative samples of the material shall be drawn as prescribed in 'Indian Standard methods for sampling of pesticides and their formulations' ( *under preparation* ).

**NOTE** — Till such time the standard under preparation is published, the samples shall be drawn as agreed to between the parties concerned.

## 5. TESTS

**5.1** Tests shall be carried out by appropriate methods referred to in col 4 and 5 of Table 1.

**5.2 Quality of Reagents** — Unless specified otherwise, pure chemicals and distilled water ( *see* IS : 1070-1977\* ) shall be employed in the tests.

**NOTE** — 'Pure chemicals' shall mean chemicals that do not contain impurities which affect the results of analysis.

# APPENDIX A

## [ Table 1, Item (i) ]

### DETERMINATION OF ETHION CONTENT

#### A-0. GENERAL

**A-0.1** For the determination of ethion content, two methods namely infra-red spectrophotometric method ( *see* A-1 ) and column chromatography followed by phosphate determination as quinoline phosphomolybdate ( *see* A-2 ) have been specified. Either of the two methods may be used for the determination of ethion content but infra-red spectrophotometric method shall be the referee method.

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\*Specification for water for general laboratory use ( *second revision* ).

## A-1. INFRA-RED SPECTROPHOTOMETRIC METHOD

**A-1.1 Method** — The method consists of dissolving ethion in tetrachloroethylene, spectroscopic grade; scanning the solution in the wave length region of  $9.5\text{--}10.5\mu$  ( $1050\text{--}950\text{ cm}^{-1}$ ) in an infra-red spectrophotometer and measuring the characteristic absorption peak at  $9.88\mu$  ( $1012\text{ cm}^{-1}$ ) through the base point at  $10.15\mu$  ( $985\text{ cm}^{-1}$ ).

**NOTE** — Since compounds with ethoxy groups attached to phosphorus exhibit strong absorption band between  $1030$  and  $1010\text{ cm}^{-1}$  and the absorption band is attributed to P-O(C) vibration in the molecule, a wide range spectrum for identification of individual peaks may be taken.

### A-1.2 Apparatus

**A-1.2.1 Infra-Red Spectrophotometer** — Capable of recording in the region of  $2000$  to  $400\text{ cm}^{-1}$ . The slit width, gain, response time and scanning speed should be so adjusted as to produce a satisfactory signal to noise ratio and an adequate resolution. For the determination of ethion content by this method following parameters have been observed to be suitable:

Slit programme	1 000
Gain	4
Attenuator speed	1 100
Scan time	$100\text{ cm}^{-1}/\text{min}$
Suppression	5
Scale	IX
Source current	0.3 A

**A-1.2.2 Absorption Cells** — sealed absorption cells with sodium chloride or potassium bromide windows having an internal light path  $0.1\text{ mm}$ .

**A-1.2.3 Hypodermic Glass Syringe** —  $1.0$  or  $2.0\text{-ml}$  capacity, fitted with an  $1.25\text{ mm}$  (18 gauge) slip on type needle.

**A-1.2.4 Volumetric Flasks** — capacity  $25\text{-ml}$ , glass stoppered.

### A-1.3 Reagents

**A-1.3.1 Ethion** — of known (plus 99 percent) purity.

**A-1.3.2 Tetrachloroethylene** — spectroscopic grade.

### A-1.4 Procedure

**A-1.4.1 Preparation of calibration graph**

**A-1.4.1.1** Weigh accurately into separate 25-ml volumetric flasks 50, 75, 100, 125 and 150 mg of pure ethion. Dissolve in tetrachloroethylene and dilute to the mark after mixing thoroughly.

**A-1.4.1.2** Fill the absorption cell with each of the five standard solutions of ethion ( *see* A-1.4.1.1 ) in turn, starting with the dilute standard solution, by means of hypodermic syringe. Make three scans of each of these solutions in the wavelength region  $9.5-10.5\mu$  (  $1050$  to  $950\text{ cm}^{-1}$  ).

**A-1.4.1.3** For each of the scans of the standard solutions, measure the absorbance at  $1012\text{ cm}^{-1}$  and the base point optical absorbance at  $985\text{ cm}^{-1}$ . The difference between the maximum absorbance at  $1012\text{ cm}^{-1}$  and the base point optical absorbance at  $985\text{ cm}^{-1}$  gives absorbance due to ethion.

**A-1.4.1.4** Average the absorbance of the three scans of each of the standard solution ( *see* A-1.4.1.1 ) and plot a graph, taking absorbance as ordinate against concentration as abscissa, for the five calibration solutions.

#### **A-1.4.2 Estimation of Ethion Content**

**A-1.4.2.1** Weigh accurately a quantity of the sample containing 100-150 mg of ethion in a 25-ml volumetric flask and proceed as given in A-1.4.1.1 to A-1.4.1.3.

#### **A-1.5 Calculation**

$$\text{Ethion content, percent by mass} = \frac{M \times 100}{M_1}$$

where

$M$  = mass, in mg, of ethion as calculated from the calibration curve.

$M_1$  = mass, in mg, of sample taken for test.

### **A-2. COLUMN CHROMATOGRAPHY/GRAVIMETRIC METHOD**

**A-2.1 Method** — Ethion is separated from impurities by column chromatography and an aliquot of purified ethion made to volume is decomposed with a mixture of sulphuric acid and nitric acid. The phosphate thus obtained is precipitated as quinoline phosphomolybdate and is estimated gravimetrically.

#### **A-2.2 Apparatus**

**A-2.2.1 Chromatographic Column** — about 40 cms in length and having 20 mm internal diameter, with a fritted glass disc ( G 3 ) at the bottom that supports the absorbant and a burette tap. The top shall have a B 19 joint fused.

**A-2.2.2 Sintered Glass Crucible** — porosity G-4, 30-ml capacity.

### **A-2.3 Reagents**

**A-2.3.1 Silica Gel** — Chromatographic grade ( 60-120 mesh ), dried for 4 hours at 110°C.

**A-2.3.2 Ether-Hexane Mixture** — 10 percent ether in hexane ( v/v ).

**A-2.3.3 Sulphuric Acid** — analytical reagent grade.

**A-2.3.4 Nitric Acid** — analytical reagent grade.

**A-2.3.5 Citric Molybdate Solution** — prepared as given in A-2.3.5.1 and A-2.3.5.2.

**A-2.3.5.1** Weigh 54 g of molybdic anhydride (  $\text{MoO}_3$  ) in a 500-ml beaker. Add 200 ml of distilled water and 11 g of sodium hydroxide pellets. Heat the beaker until molybdic anhydride dissolves.

**A-2.3.5.2** Dissolve 60 g of citric acid crystals in 250 ml of water in a one litre beaker and add 140 ml of hydrochloric acid (relative density 1.16). To this solution, add solution prepared in A-2.3.5.1 with stirring. Filter and dilute to one litre. This solution may be green or blue in colour. Add 0.5 to 1 percent solution of potassium bromide until green colour disappears. Keep solution in a dark place in a well stoppered bottle of polyethylene.

**A-2.3.6 Quinoline Solution** — dilute 60 ml of concentrated hydrochloric acid with 300-400 ml water in a beaker and heat to 70 to 80°C. Slowly add 50 ml of pure quinoline stirring all the while when quinoline dissolves. Cool the solution, dilute with distilled water and filter into a one litre volumetric flask and make up the volume with water. Keep solution in a polyethylene bottle.

### **A-2.4 Procedure**

**A-2.4.1** Weigh about 22.5 g of silica gel at room temperature into a 250-ml beaker and make a slurry with hexane. Mix the slurry with a glass rod and transfer to the column by keeping the stop-cock open and without allowing the gel to dry during the operation by using more hexane. Gently tap the column to drive out any air bubbles entrapped. Apply nitrogen pressure 0.7 Kg/cm<sup>2</sup> ( 10 lbs ) on the column for compaction of the gel. Do not allow the liquid level to drop below the top of the silica gel column. Layer about 2 cms of washed sand on the top of the column to prevent disturbing the silica gel while charging the solvent or sample.

**A-2.4.2** Accurately weigh a sample containing 0.5 g of ethion technical in a 25-ml beaker and dissolve in about 2-3 ml of hexane. When the liquid level is about 1 mm above the silica gel, pour the solution of ethion

carefully along the sides of the column. Rinse the beaker and the column with three times 2 ml of hexane. Again, do not allow the liquid level to fall below the top of the silica gel column.

**A-2.4.3** Elute with 250 ml of 10 percent ether in hexane at a rate of 5 ml/min. Discard the first 50 ml fraction and collect the next 150 ml fraction in a cylinder. Transfer this fraction into a 250-ml volumetric flask quantitatively using hexane and make up to the mark.

**A-2.4.4** Pipette out 25 ml aliquot (*see* A-2.4.3) into a dry Kjeldahl flask of 100-ml capacity. Evaporate the solvent on a water bath. Cool the flask and add carefully 4.5 ml of concentrated sulphuric acid and 1.5 ml concentrated nitric acid. Heat on a low flame till dense fumes appear. Let it cool and repeat the operation. Add 1 ml nitric acid and continue heating till the solution becomes perfectly clear and almost colourless.

**A-2.4.5** Transfer the contents quantitatively, on cooling, into a 500-ml beaker with water to a volume of about 25-30 ml. Boil for 5 minutes and dilute with water to about 150 ml. Add 25 ml of citric molybdate solution and boil for 3 minutes. Immediately add 12.5 ml of quinoline solution from a burette. The quinoline solution should be added dropwise and the contents of the beaker must be stirred continuously so that the precipitate obtained can be easily filtered. Let the beaker stand for 5 minutes stirring for 5-10 seconds each time and then cool it to room temperature.

**A-2.4.6** Filter by suction through a G4 sintered glass crucible. Wash precipitate four times by decantation and then wash with plenty of distilled water. Dry the precipitate at 110°C for 2 hours.

**A-2.4.7** Weigh the precipitate as quinoline phosphomolybdate  $(C_9H_7N)_3H_3PO_4 \cdot 12 MoO_3$ .

## A-2.5 Calculation

Ethion content, percent by mass

$$= \frac{384.5}{2 \times 31} \times 0.01399 \times \frac{m}{M} \times \frac{250}{25} \times 100$$

$$= \frac{86.76 \times m}{M}$$

where

$m$  = mass, in g, of the precipitate obtained; and

$M$  = mass, in g, of the sample taken for test.

**NOTE** — 0.01399 is the conversion factor for quinoline phosphomolybdate to phosphorus as P.

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*( Continued from page 2 )*

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